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Answers

11. molar mass = mass/moles; mass = 45.0 g; find moles; $T_b = 90.2^\circ\text{C} - 80.1^\circ\text{C} = 10.1^\circ\text{C}$

$$\text{moles} = \frac{\Delta T_f}{2.53} \times \text{kg benzene}$$
$$= \frac{10.1^\circ\text{C}}{2.53} \times 0.0750 = 0.300 \text{ moles}$$

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Ans: Primary alcohols: (i), (ii), (iii)

Secondary alcohols: (iv), (v) Tertiary

alcohols: (vi) 11.2. Identify allylic

alcohols in the above examples. Ans: (ii)

and (iv) i.e. $\text{H}_2\text{C}=\text{CH}-\text{CH}_2\text{OH}$ and.

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Answers
and 2-methyl magnesium iodide (c)
2-dimethylpropanone and methyl
magnesium iodide (d) 2-methylpropanal
and isopropyl magnesium iodide. Answer.
Answer: (d) 2-methylpropanal and
isopropyl magnesium iodide

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