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#### Vocabulary Section

- 13.1 1. electric current 2. electrical symbols 3. switch 4. open circuit 5. closed circuit 6. resistance
- Section 13.2 7. ampere 8. battery 9. voltage 10. volt
- Section 13.3 11. ohm 12. Ohm ' s law 13. resistance 14.

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ANSWER Answer the following questions in the space provided. 1. Following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid. b.

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13.2) Draw the line bond structure for the general form of an amino acid using “ R ” to represent the side chain. Draw

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the N- and C-terminus groups using the forms that are predominant at physiological pH (pH = 7.4). Include lone pairs in your drawing.

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symbols 3. switch 4.  
open circuit 5. closed  
circuit 6. resistance

Section 13.2 7.  
ampere 8. battery 9.  
voltage 10. volt

Section 13.3 11. ohm  
12.

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ANSWER Answer the following questions in the space provided. 1.

Following are statements about the dissolving process.

Explain each one at the molecular level. a.

Increasing the

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pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid. b.

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#### Solutions

SHORT ANSWER 1. a. Increasing the pressure of the solute gas above the solution puts stress on the equilibrium of

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the system. Gas molecules collide with the liquid surface more often, causing an increase in the rate of gas molecules entering into solution.

b. As the temperature of the water increases, water molecules move faster,

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### Ions in Aqueous Solutions and Colligative Properties

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PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

1. \_\_\_\_\_ a.  
Predict the boiling point of a 0.200 m solution of glucose in water.

\_\_\_\_\_ b.  
Predict the boiling

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point of a 0.200 m  
solution of potassium  
iodide in water. 2.

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